

Coating process of solid dosage forms

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Aims of coating

- to mask the taste, odour or colour of the drug**
- to provide physical and chemical protection for the drug**
- to separate incompatible ingredients**
- to control the release of the drug from the dosage form**
- to give an elegant finish to the tablet**

Requirements of coating layer

- its surface is faultless, smooth, polished and uniformly;**
- it has appropriate mechanical hardness;**
- it protects the tablet core from the air, moisture and light;**
- it masks the unpleasent taste perfectly;**
- it dissolves rapidly in the gastric or intestinal juice, when it is necessary;**
- it is as thin as possible;**
- when the coating layer contains drug, it must be compatible.**

Types of coating process

1. Sugar coating

- subcoating
- smoothing
- colouring
- polishing

2. Film coating

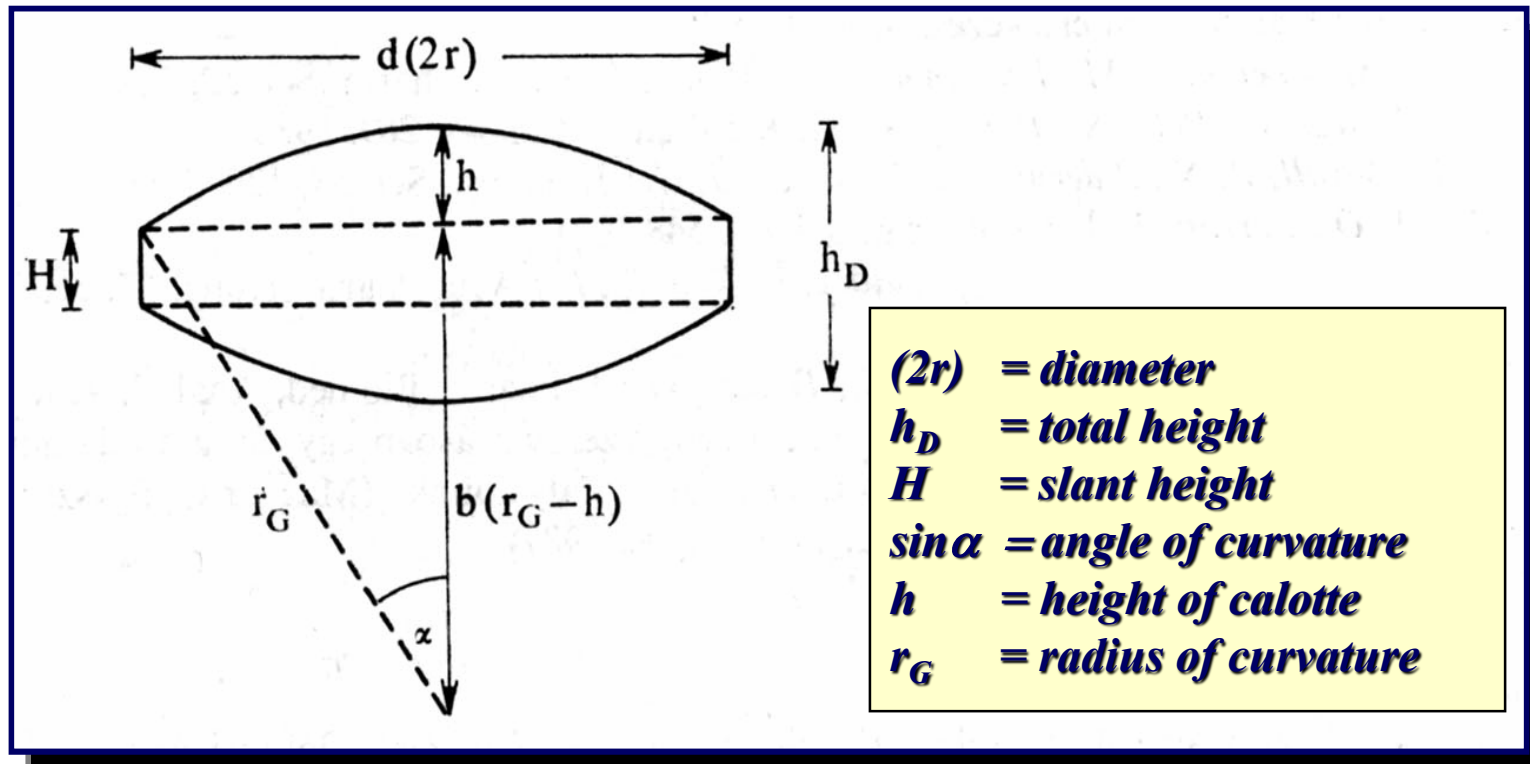
- gastric coated
- enteric coated
- permeable coating

3. Melted coating

4. Dry coating

5. Electrostatic coating

Characteristical parameters of tablet core



The important physiological factors

- a) the length of time of passage**
- b) the role of pH**
- c) the effects of enzymes**

The pH in the GI tract

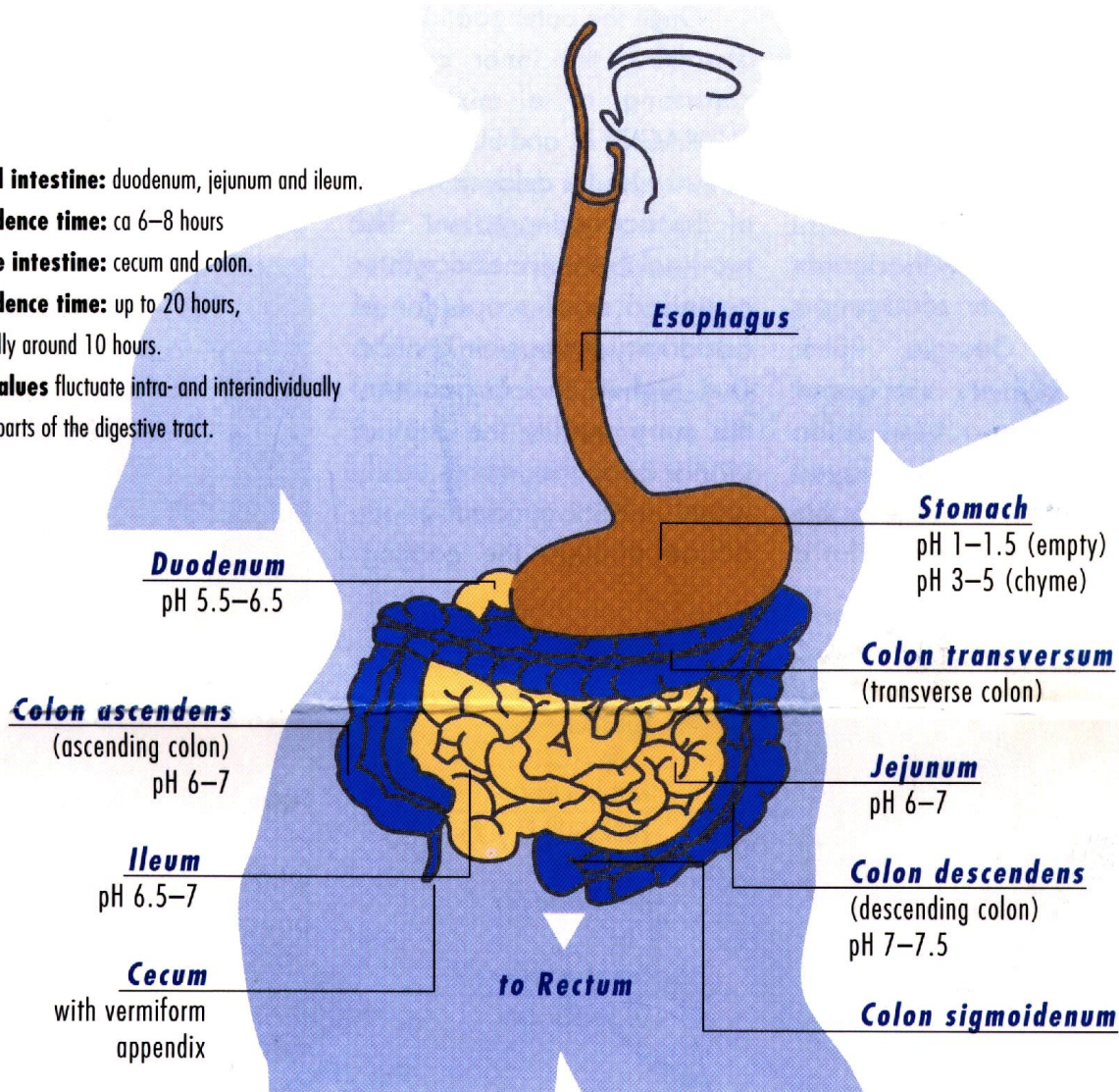
Small intestine: duodenum, jejunum and ileum.

Residence time: ca 6–8 hours

Large intestine: cecum and colon.

Residence time: up to 20 hours,
typically around 10 hours.

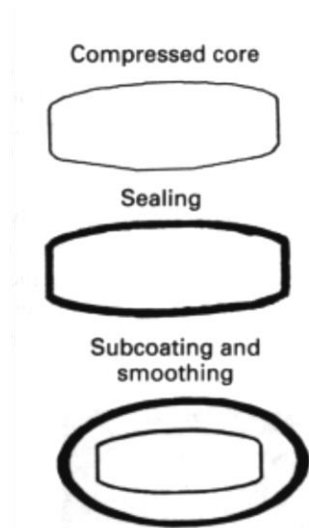
pH values fluctuate intra- and interindividually
in all parts of the digestive tract.



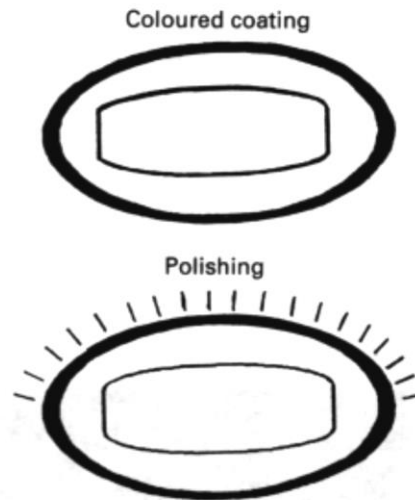
Sugar coating

Phases of sugar coating

Subcoating



Colouring



Smoothing

Polishing

Groups of colours

- 1. Synthetic colours**
- 2. Vegetable and animal colours**
- 3. Nature pigments**
- 4. Synthetic pigments**
- 5. Lacker materials**

Requirements

- **lastingness of colour**
- **intense colouring effect**
- **the colours of different production series must be identical**
- **must be compatible**
- **must be resistant against atmospheric heat and moisture and production effects**
- **must be stable during a long storage period**

Film coating

Disadvantages of sugar coating

- *doubles the mass and increases the size*
- *harmful for children*
- *the sugar layer is brittle*
- *the sugar layer is not tropic-resistant*
- *the process needs long time*

Advantages of film coating

- *minimal mass increase*
- *a significant reduction in processing time*
- *increased process efficiency*
- *may be tropic-resistant*
- *pH-dependent or independent film*

Requirements of film coating materials

- *not toxic*
- *colourless, tasteless, odourless*
- *resistant against atmospheric effects*
- *chemically indifferent*
- *dissolve in the gastric or/and intestinal juices*
- *must be economical*

Film forming polymers (1)

Cellulose ethers

carboxi methyl cellulose (CMC)

*sodium or calcium carboxi methyl cellulose
(Na or CaCMC)*

ethyl cellulose (EC)

hydroxi ethyl cellulose (HEC)

hydroxi propyl cellulose (HPC)

hydroxi propyl methyl. cellulose (HPMC)

methyl cellulose (MC)

Cellulose esters

cellulose acetate phtalate (CAP)

hydroxi propyl methyl cellulose phtalate (HPMCP)

Film forming polymers (2)

Copolymers of methacrylic acid

Eudragit L 100-55 ill. L30D

Eudragit S 100

Kollicoat MAE 30DP

Amino alkyl methacrylate copolymer

Eudragit E 100

Film forming polymers (3)

Methacrylic ester copolymer

Eudragit RL 100 ill. RL 30D

Eudragit RS 100 ill. RS 30D

Eudragit NE 30D

Kollicoat EMM 30D

Polivynil acetate copolymer

Kollidon VA 64

Kollicoat SR 30D

Polivynil pyrrolidone

Kollidon

*Acryl-EZE,
Acryl-EZE MP*

Types of aqueous polymers dispersion

*Sustained release
(neutral groups)*

Ethyl cellulose

**Aquacoat ECD,
Surelease**

Methacrylic ester copolymer

Eudragit NE, RL, RS

Kollicoat EMM

Polivynil acetate copolymer

Kollicoat RS

*Delayed release
(acidic groups)*

Cellulose esters

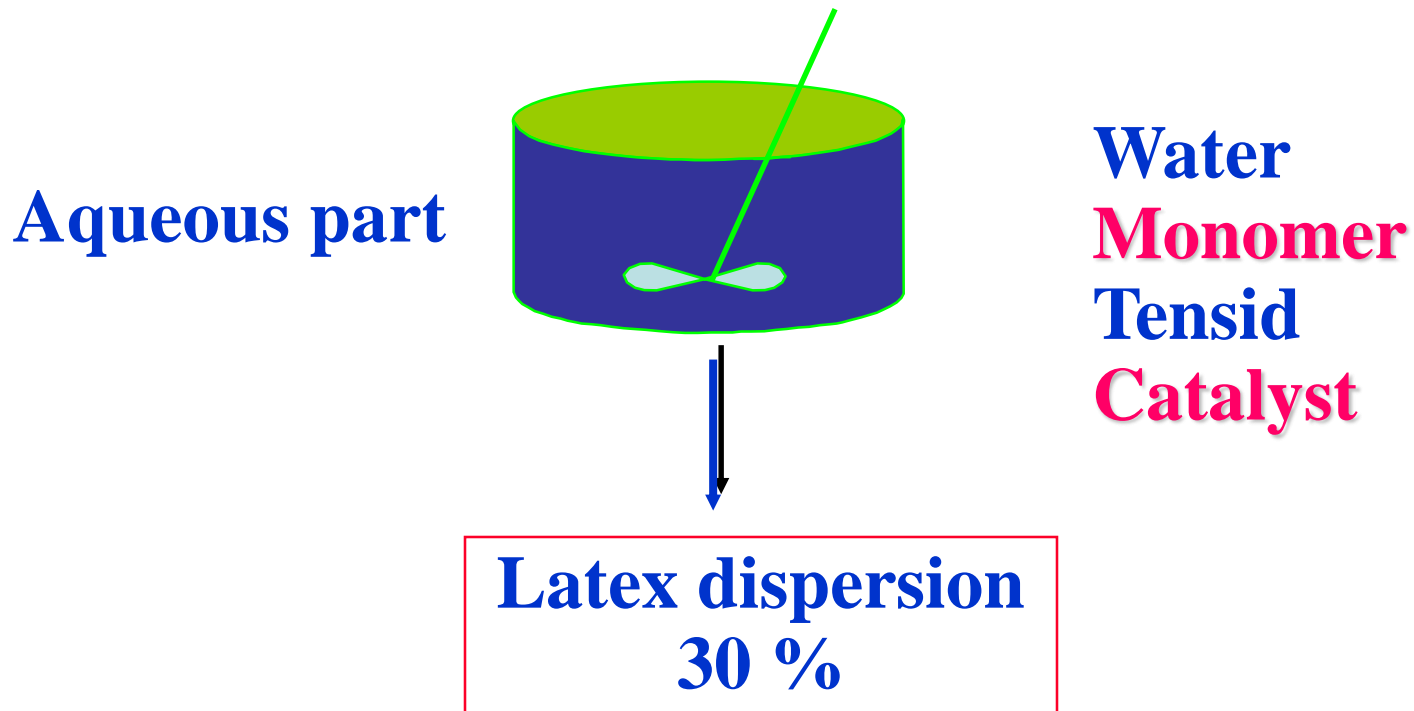
Aquacoat CPD

Acrylate copolymers

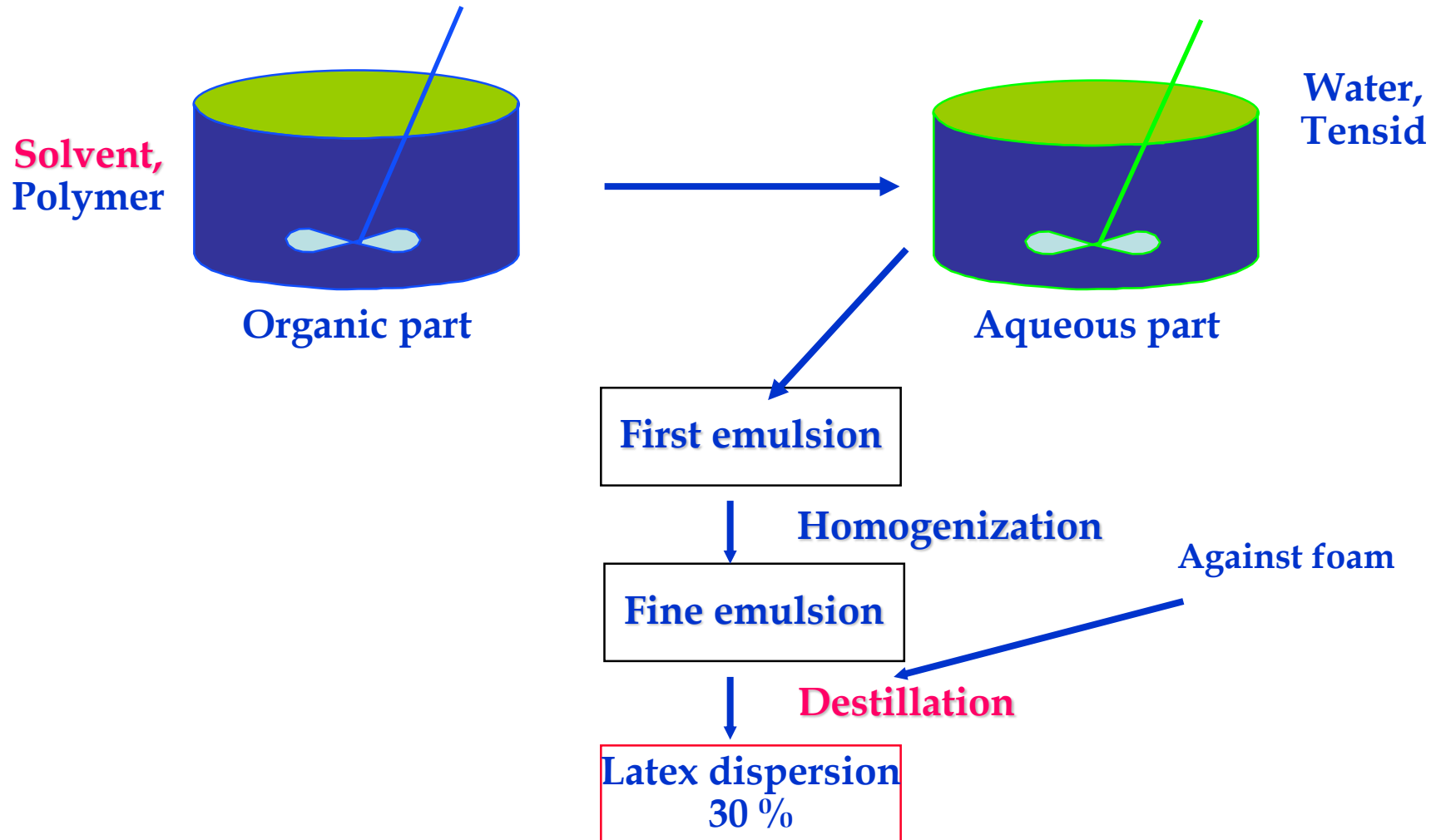
Eudragit L,

Kollicoat MAE

Process of polymerization

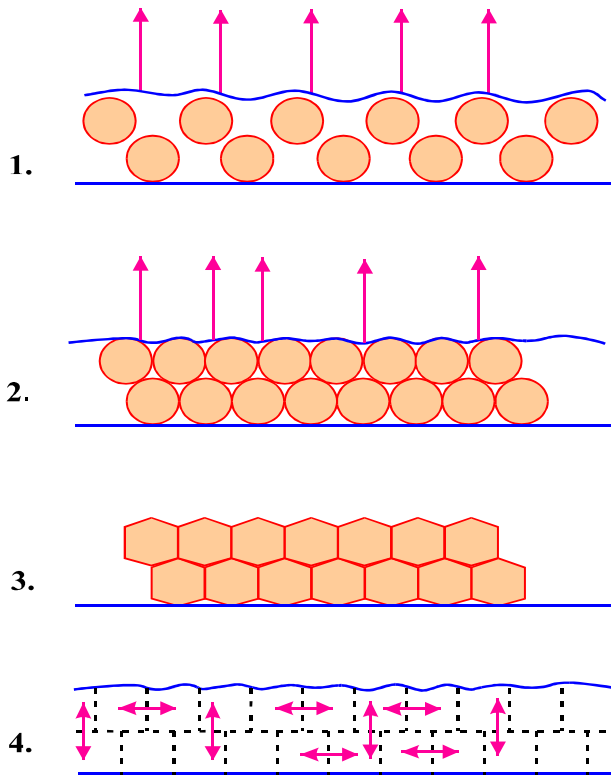


Process based on solvent evaporation



Mechanism of film forming from aqueous dispersion

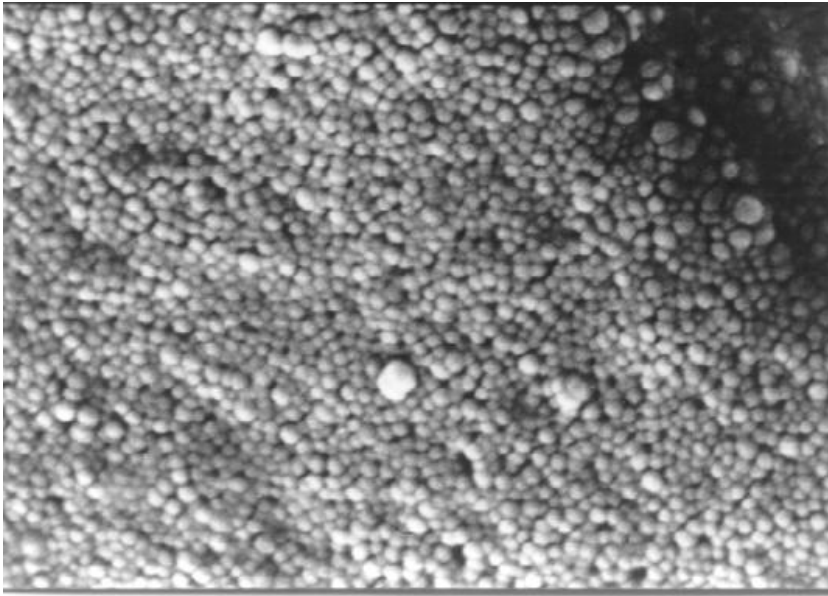
Mechanism of film forming



- 1. Water evaporation, capillary forces act between particles.*
- 2. The particles close up.*
- 3. Deformation of particles.*
- 4. Coalescence of particles.*

G. Cole, J. Hogan, M. Aulton:
Pharmaceutical Coating Technology
Taylor & Francis Ltd., 1995.

Mechanism of film forming



Starting of coalescence



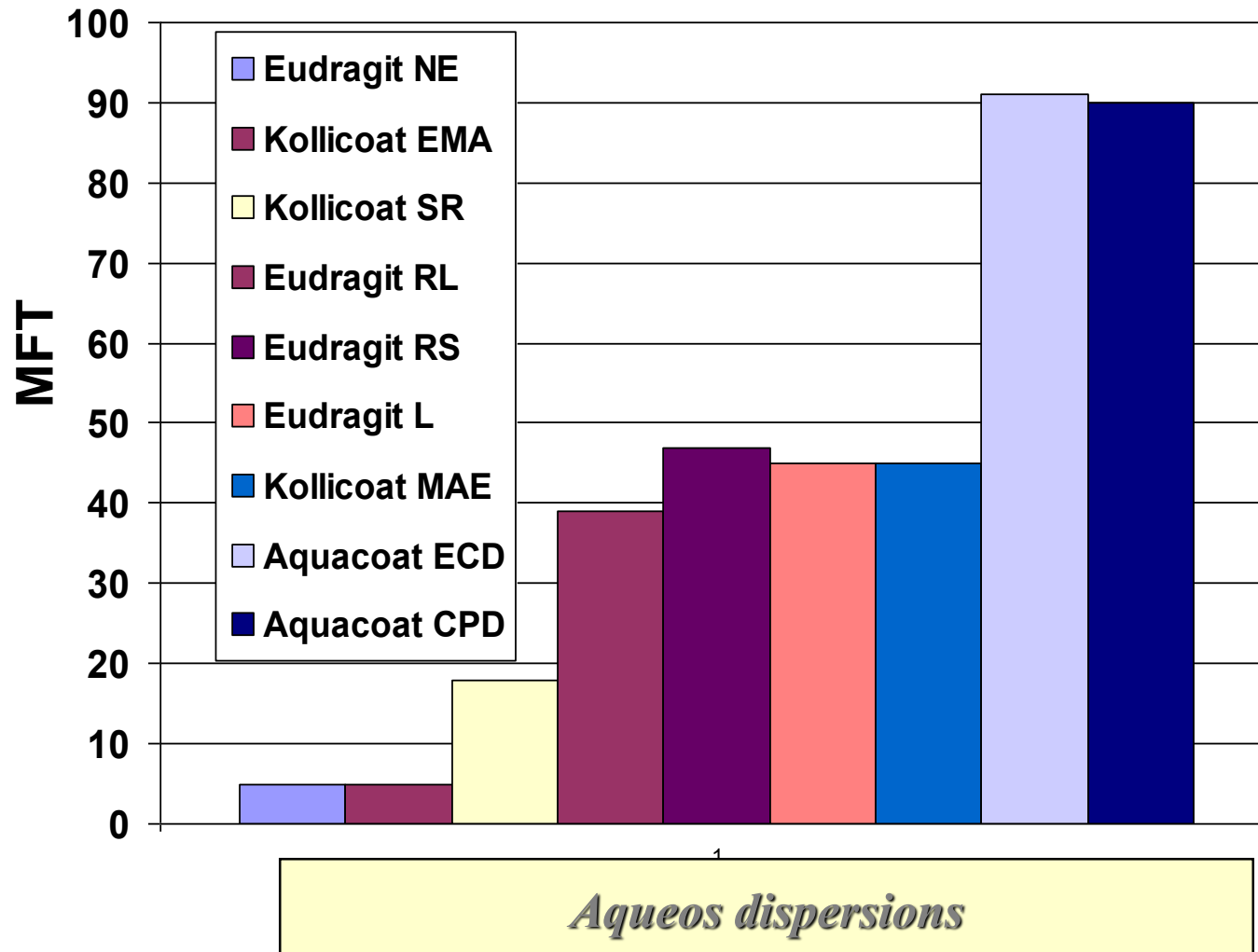
Finishing of coalescence

MFT is the key-parameter!

The temperature needs to form a homogen, trasparence film.

During coating the drying temperature must be 10-15 °C above MFT

Minimal film forming temperature (MFT)



Plasticizers

Groups

1. Polyols

- ***glycerol***
- ***propylen glycol***
- ***PEG 200-6000***

2. Organic esters

- ***phtalate esters (diethyl, dibutyl)***
- ***dibutyl sebacate***
- ***citrate esters (triethyl, acetyl-triethyl, -tributyl)***
- ***triacetin***

3. Oils/glycerids

- ***castroil oil***
- ***acetyl monoglycerids***
- ***cocoa-nut oil***

Decrease of the MFT

Plasticizers (10 -30 %)

Triethyl citrate

Dibutyl sebacate

Glyceryl triacetate (Triacetin)

Propylene glycol

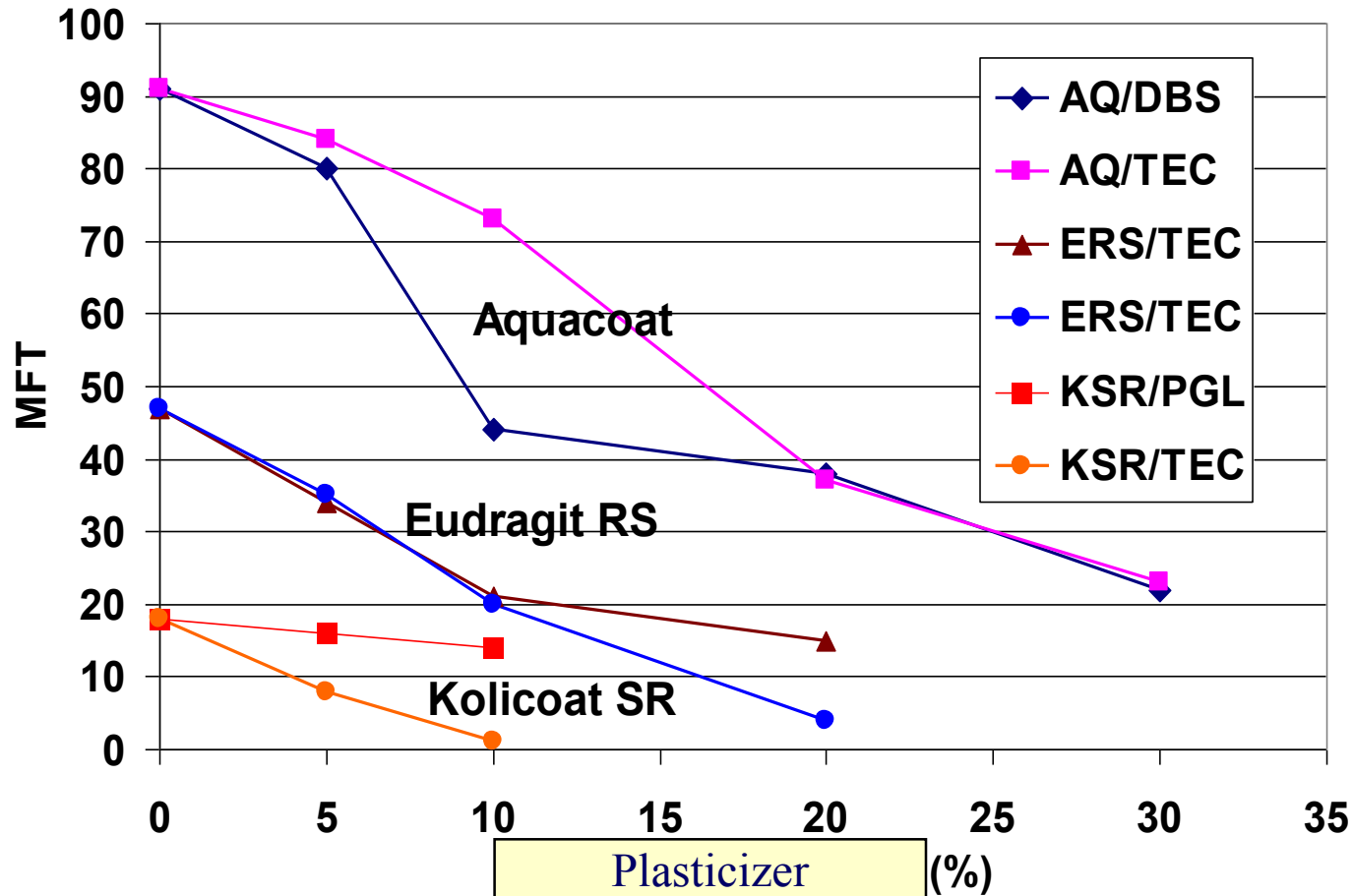
Polyaethylene glycol

Diethyl phthalate, Dibutyl phthalate

Acetyl triethyl citrate

Acetyl tributyl citrate

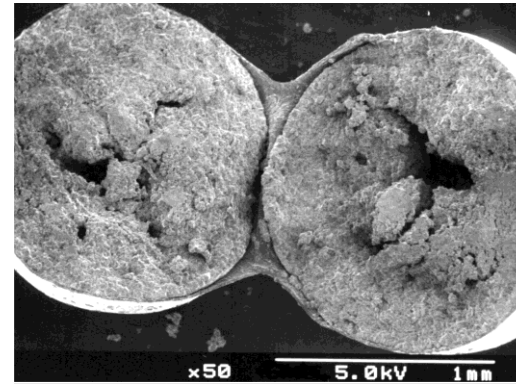
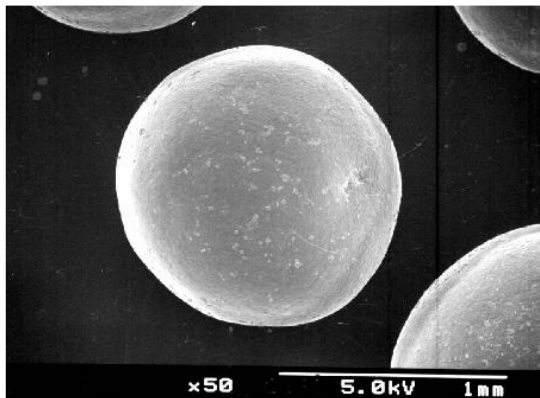
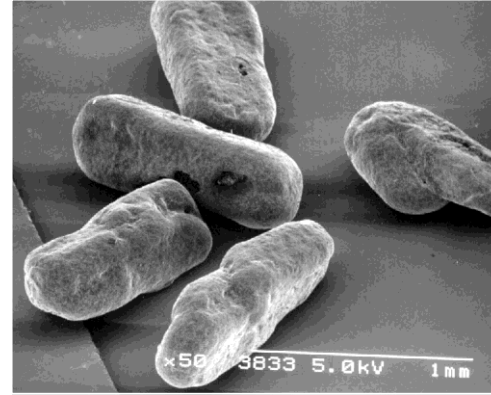
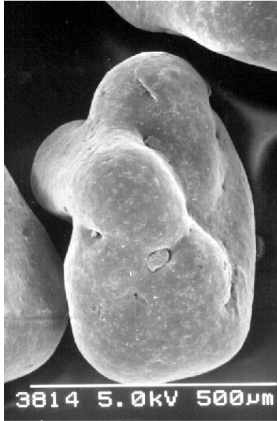
Effect of plasticizers on the MFT



Effect of plasticizers

- decrease the minimal film forming temperature
- increase the elasticity of films
- decrease the tensile strength of the films
- increase the stickiness of the film
- influence the dissolution rate

Film coated crystals, pellets



Melted coating

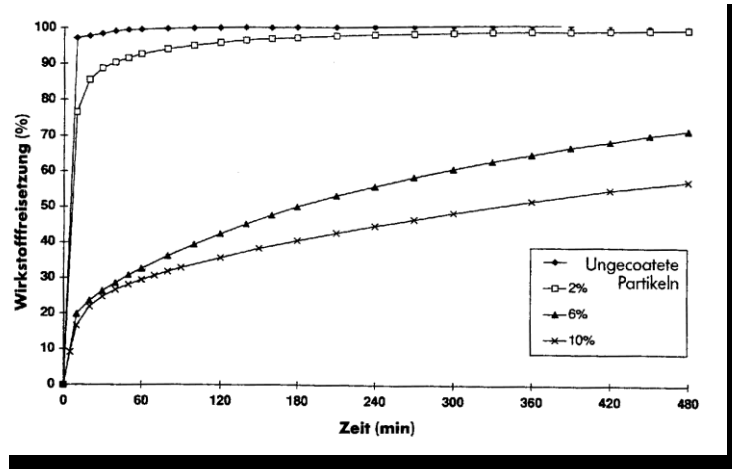
Advantages:

- *shorter process time*
- *the coating materials are used in the food industry*
- *economical*
- *the drug release may be controlled by the temperature*

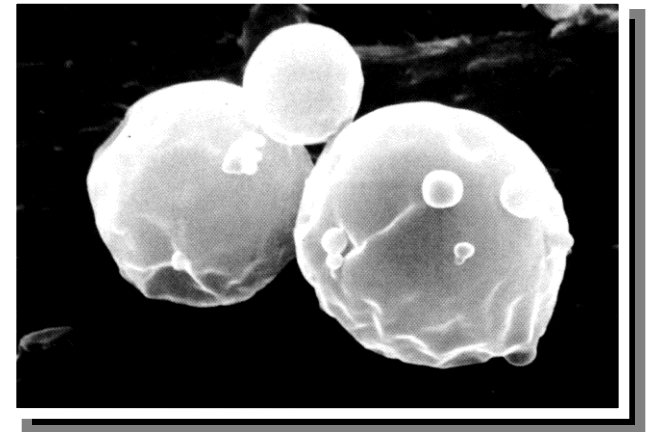
Coating materials

- *chocolate*
- *PEG 4000 és 6000*
- *waxes*
- *lipid esters*

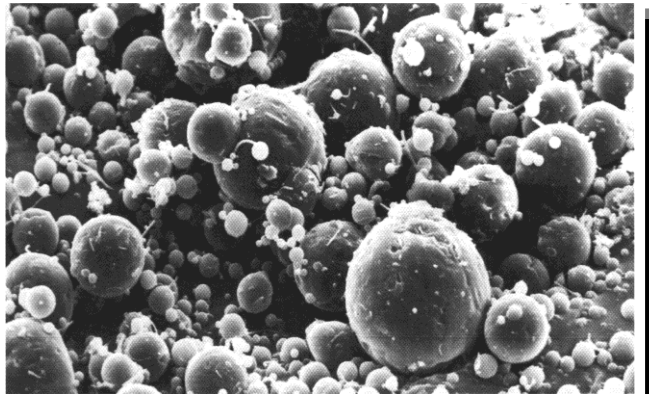
Melted coating



Release of Theophylline coating material: Compritol 888 ATO



*Precirol ATO 5
(gliceryl palmitosztearate)*

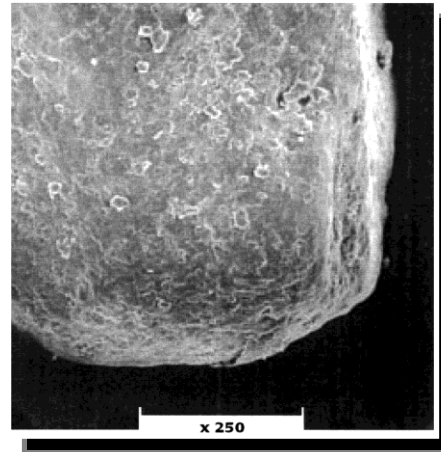
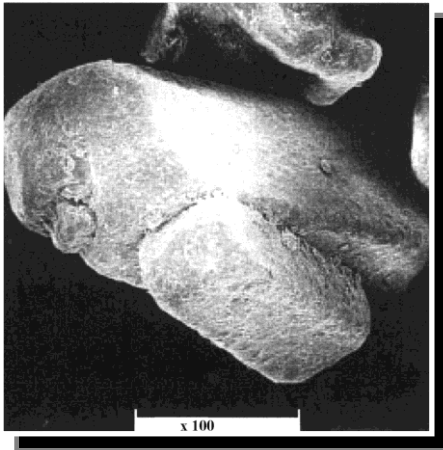


*Compritol 888 ATO
(gliceryl. behenate)*

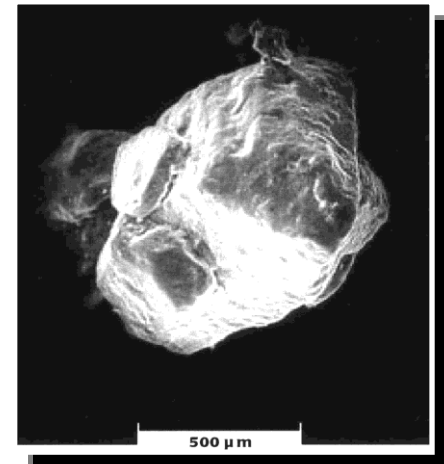
Melted coating

Gattaprine (Acetyl salicylic acid)

(gliceryl-behenate)



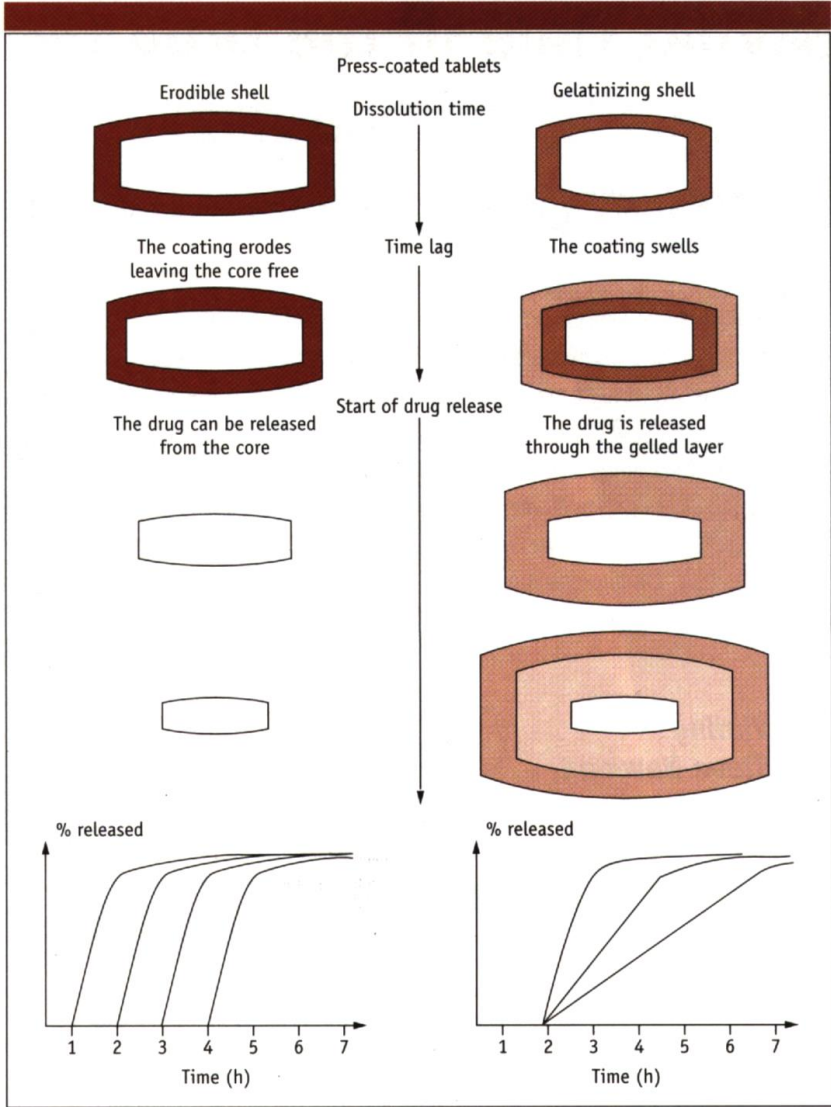
Gattaphen T (Paracetamol)



(gliceryl palmitosztearate)

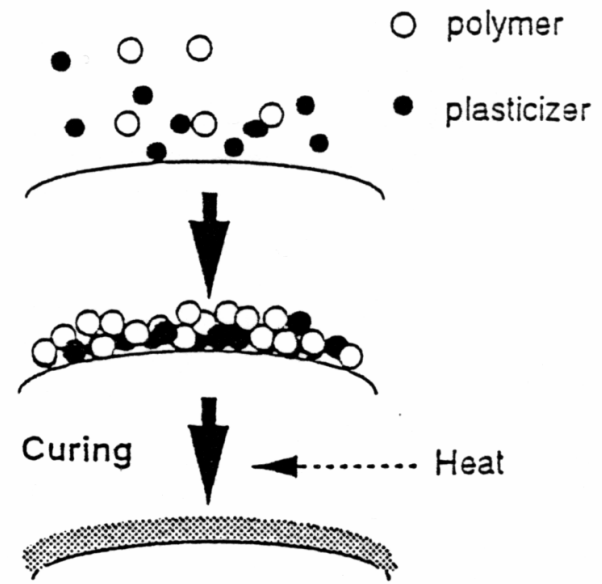
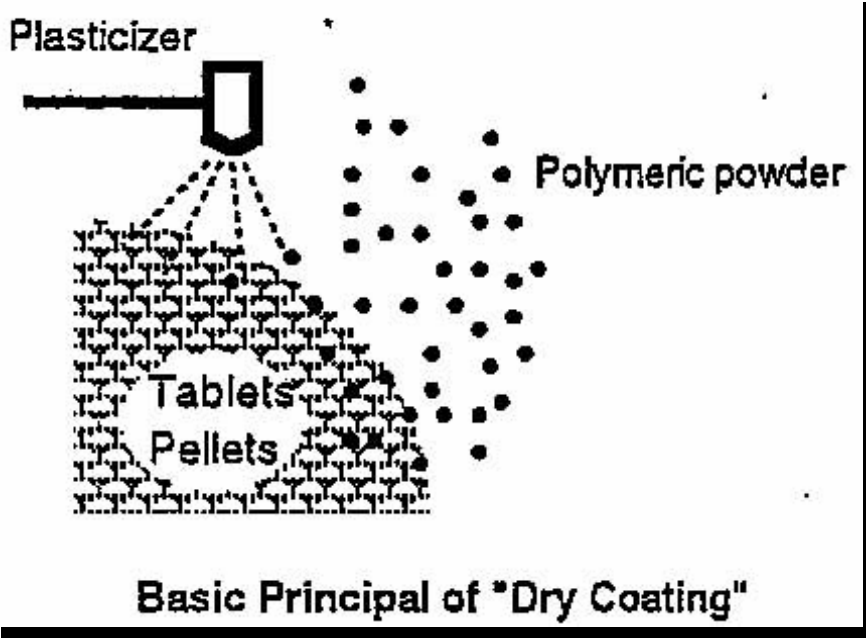
Dry coating

Compressed coating



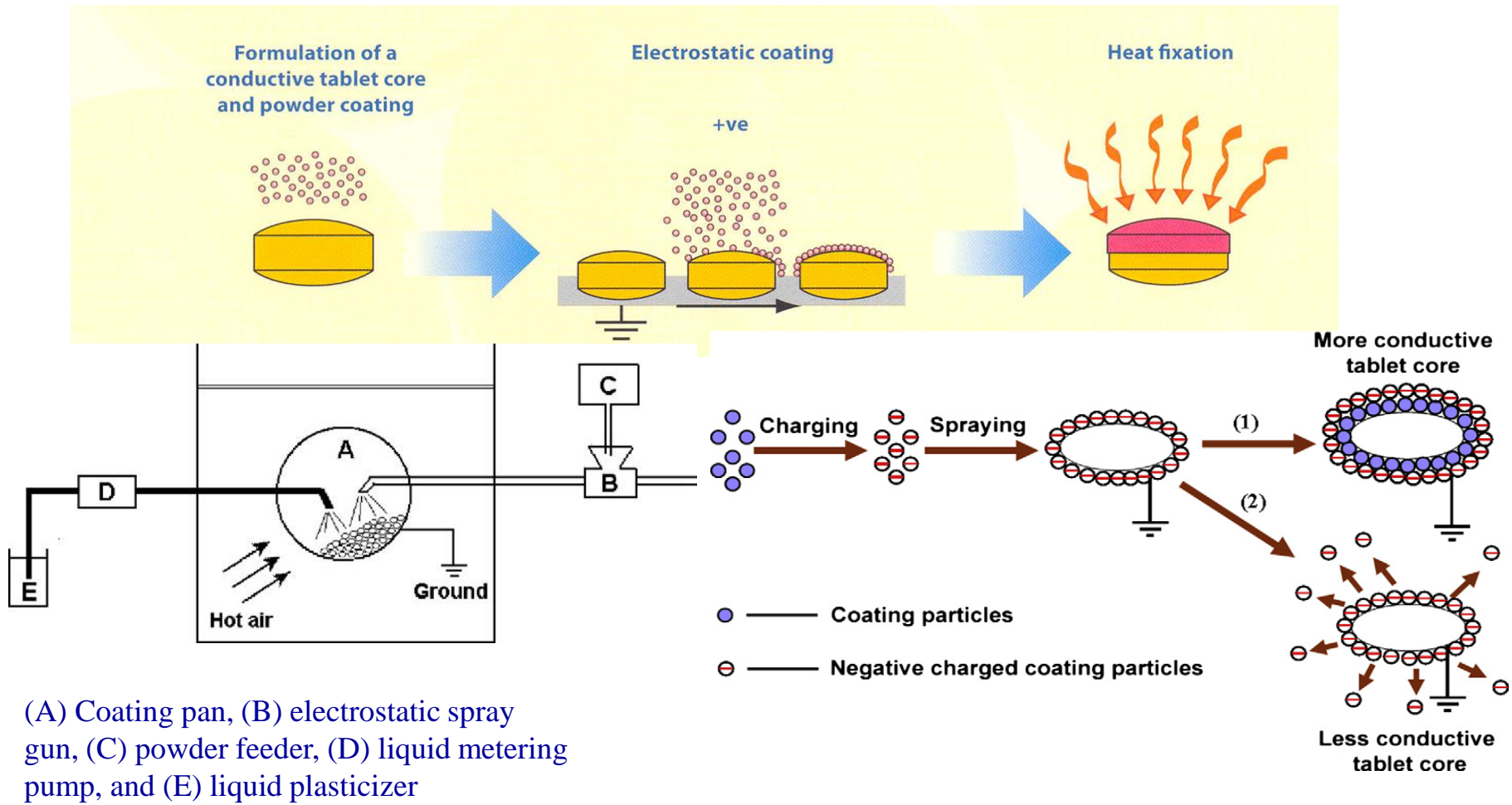
Dissolution behaviour of press-coated delivery devices.

Coating with polymer powder



Film Formation Process in "Dry Coating"

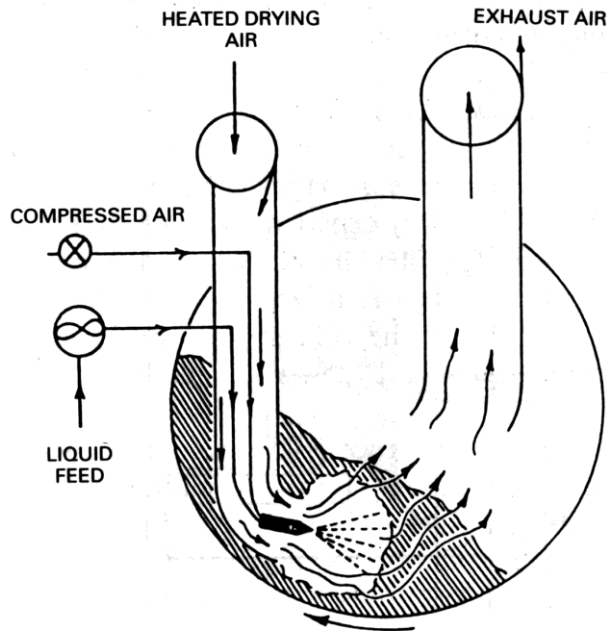
Electrostatic coating



(A) Coating pan, (B) electrostatic spray gun, (C) powder feeder, (D) liquid metering pump, and (E) liquid plasticizer

Manufacturing

Coating pan



g. 8.3 Standard coating pan using the immersion tube system.

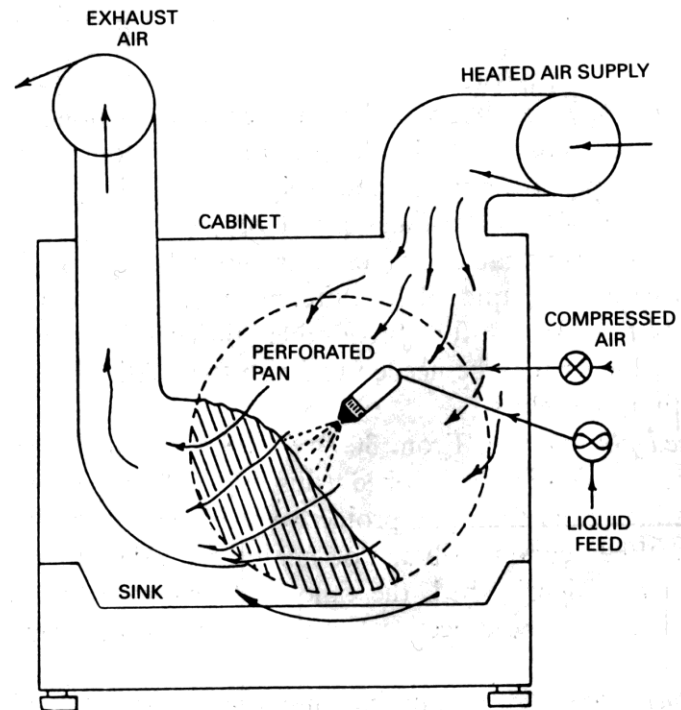


Fig. 8.4 Manesty Accelacota.

Hi-coater

Heated
drying air

Exhaust
air



Heated
drying air



Exhaust
air

Coating equipment



LAB-coater



spray gun

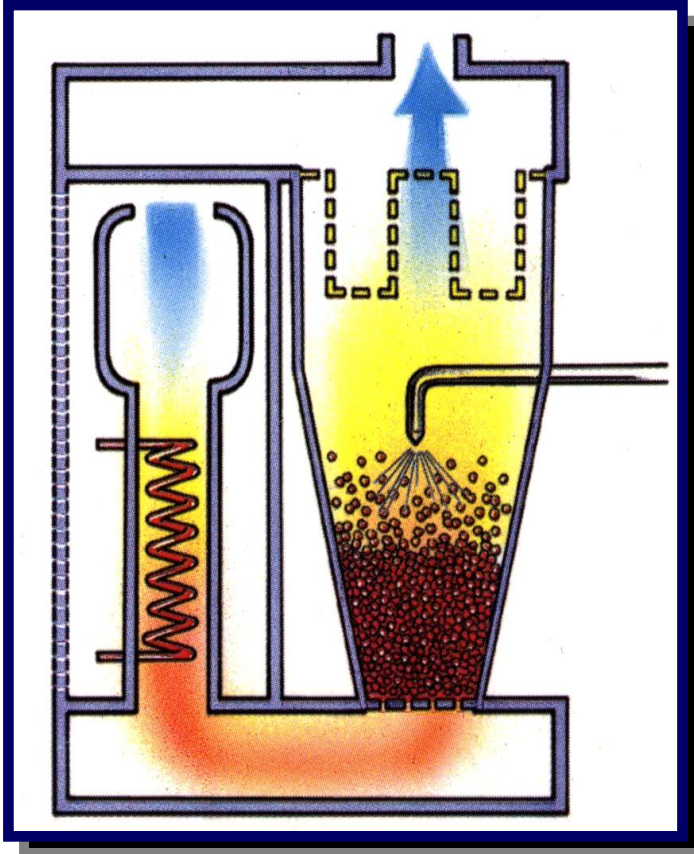
Accelacota 10 Perforated Pan Coater



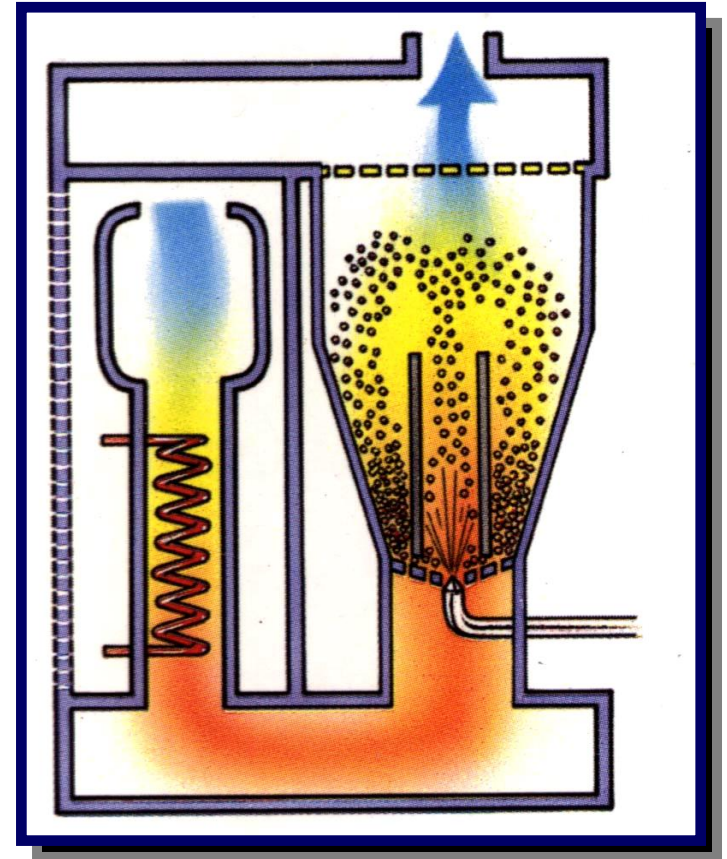
Moving Tablet Bed in Perforated Pan



Fluid bed coating

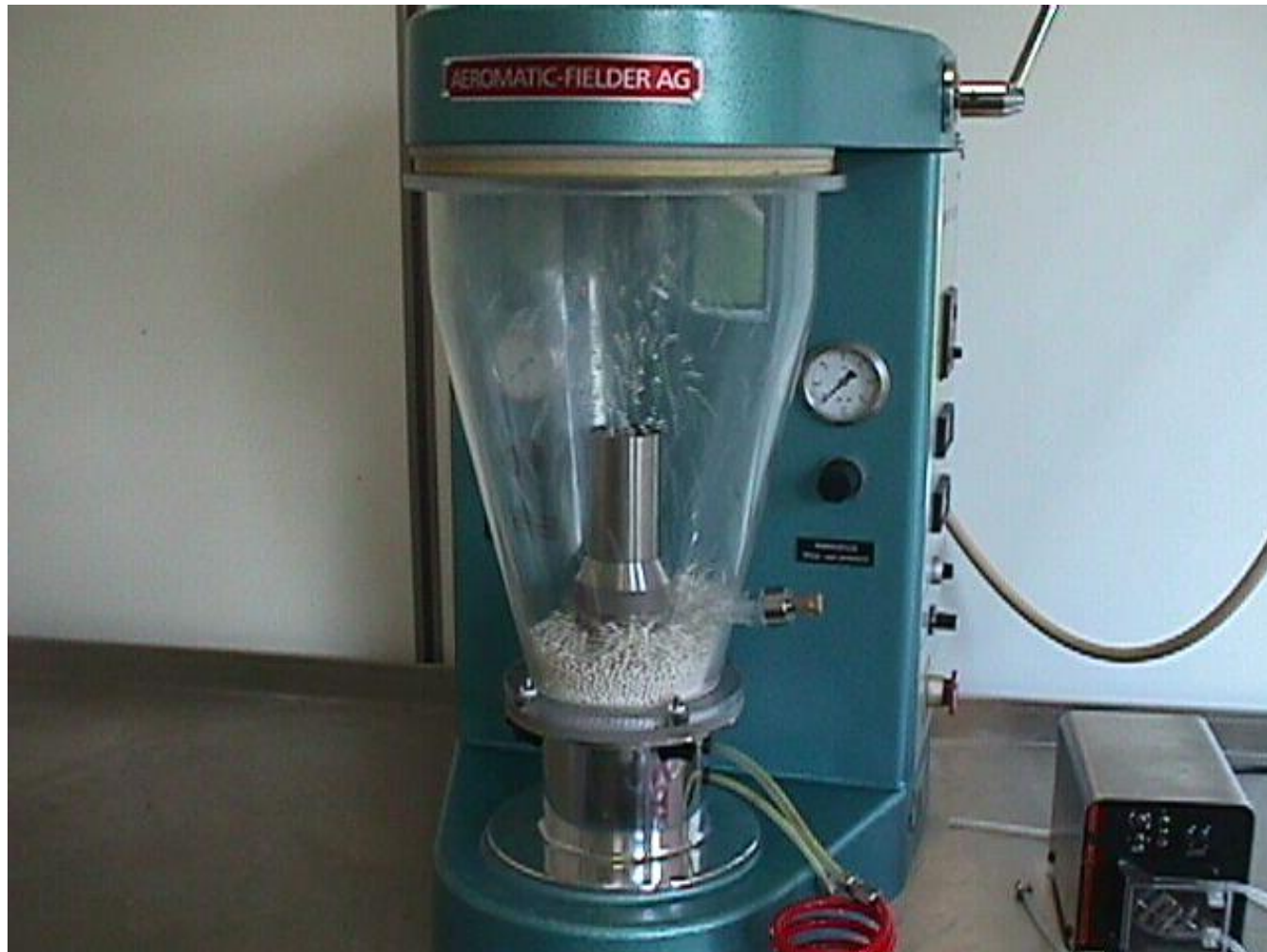


Upper spraying (Strea-1)

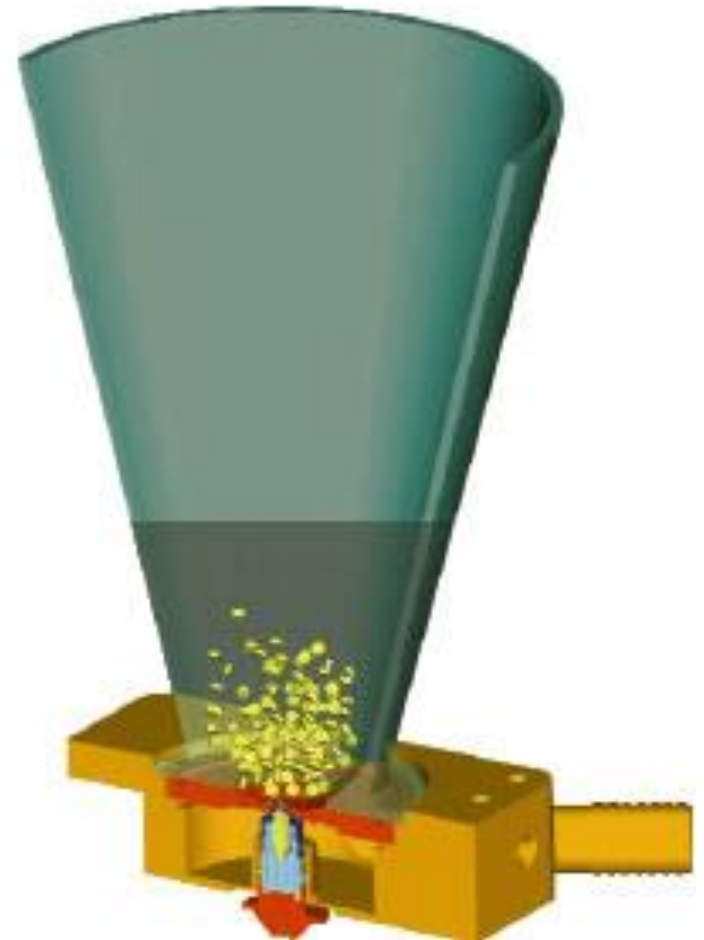
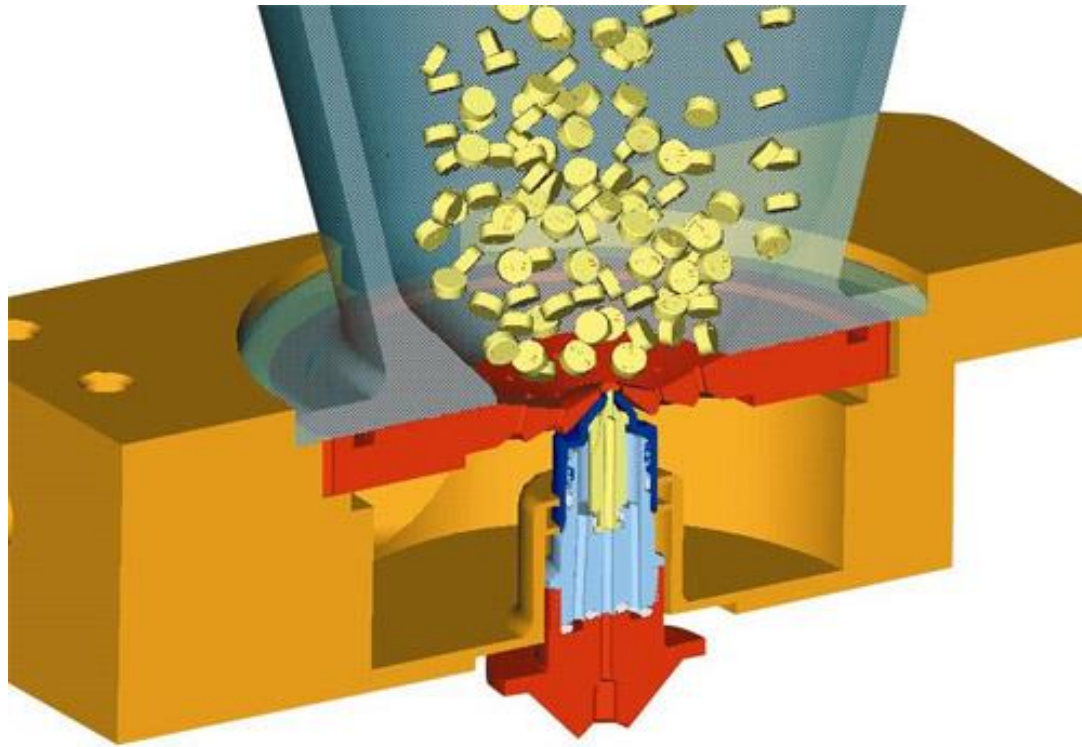


*Lower spraying (Strea-1)
(Wurster principle)*

Wurster Fluid Bed Coating



Supercell

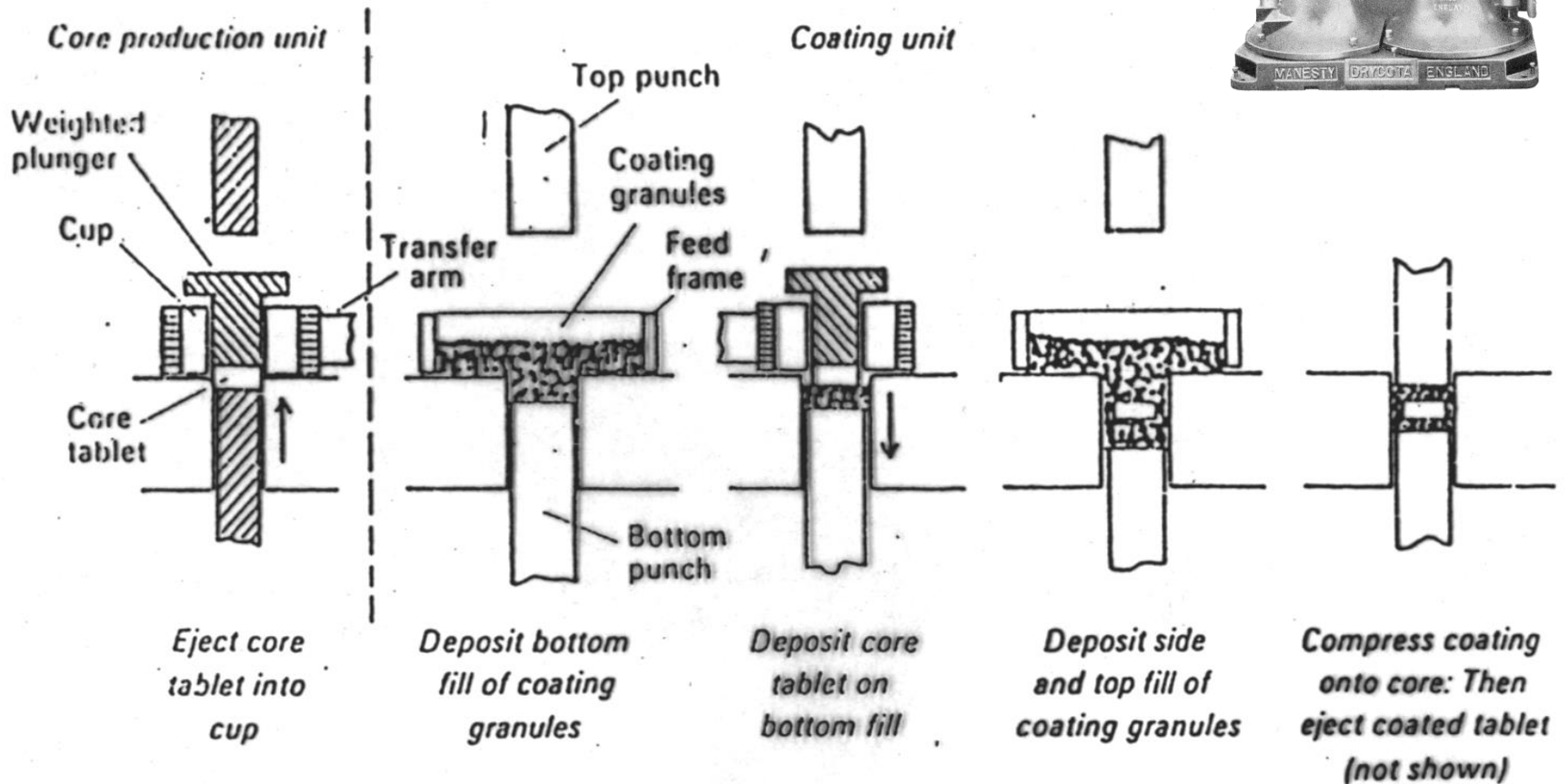
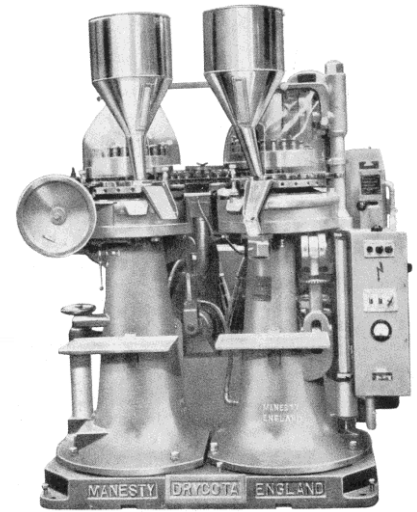


Continuous coater

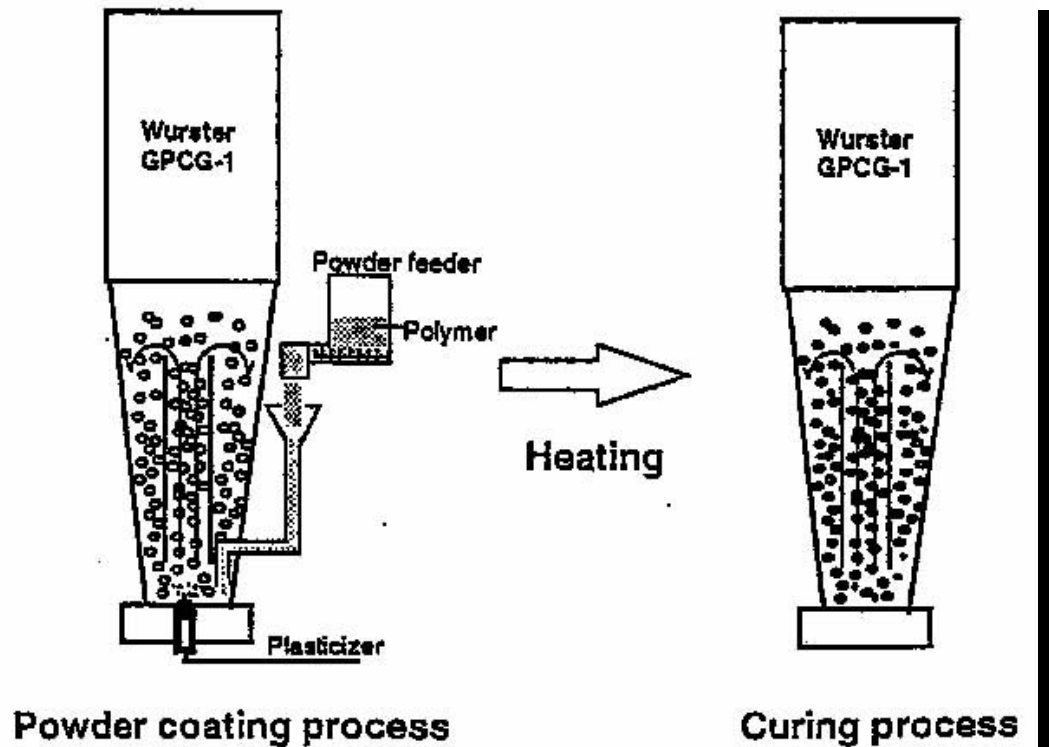
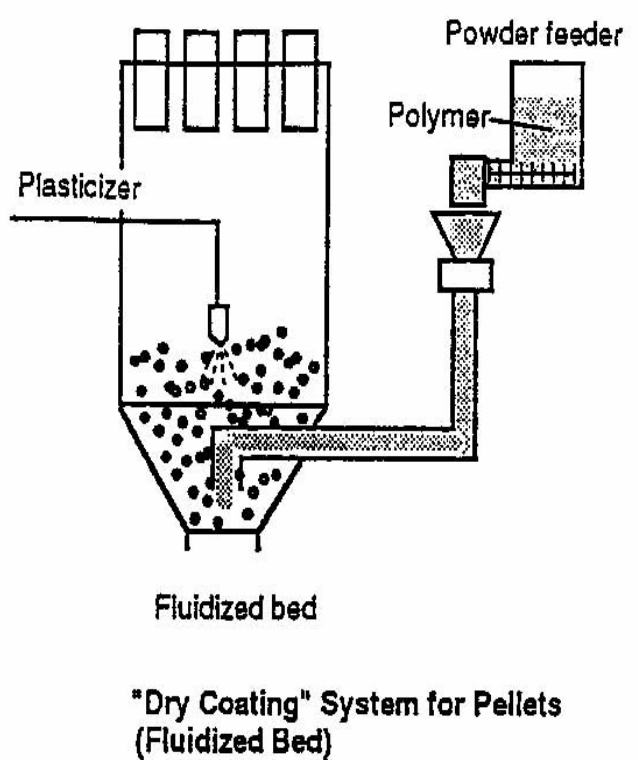
O'Hara



Compressed coating



Dry coating in fluid bed apparatus (pellet)



Tests

Aim:

- 1. Suitable for the therapy (uniformity of mass, assay, content uniformity)*
- 2. Meeting the requirements of packaging and transport (size parameters, mechanical hardness, etc.)*
- 3. The drug release is suitable (bioavailability)*

Tests (1)

1. Core

- *macroscopical test, uniformity of mass, geometry*
- *test of composition (identity, purity, assay, content uniformity)*
- *mechanical test (breaking hardness, friability)*
- *disintegration time*
- *porosity*
- *drug release*

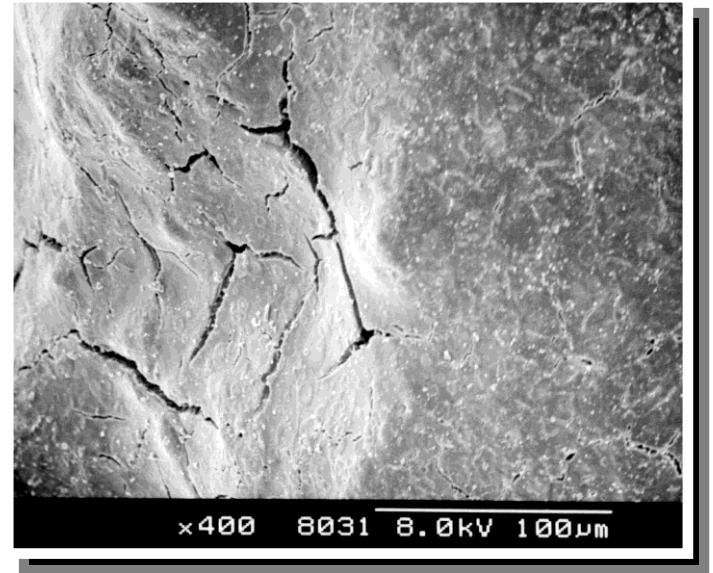
Tests (2)

- 2. The colour and glitter of coating*
 - *shade of colour, tonality, deep of colour*
 - *reflection*

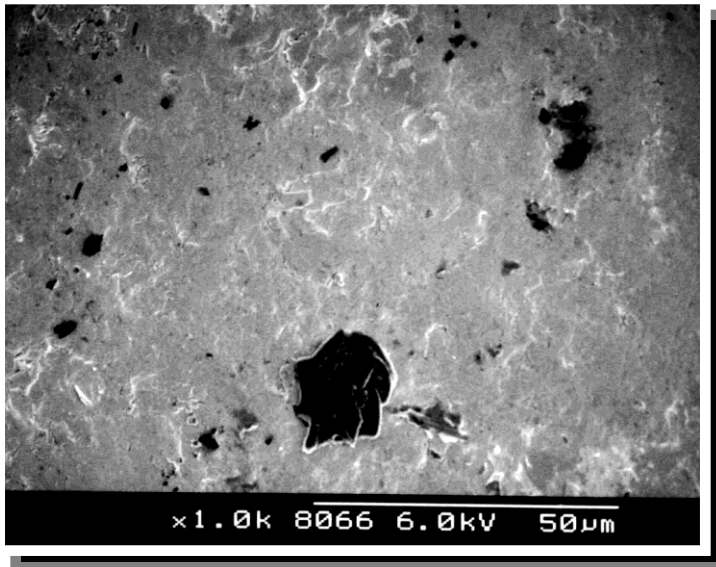
- 3. Finally coated dosage form*
 - *disintegration*
 - *drug release*

Coating defects

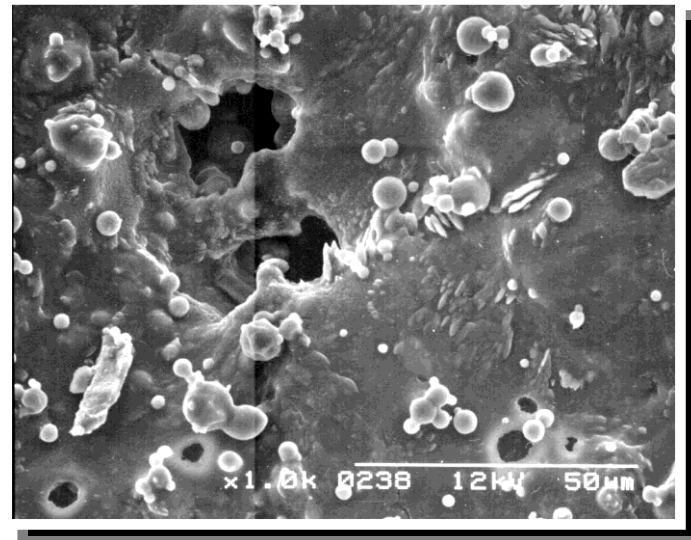
cracking



cratering

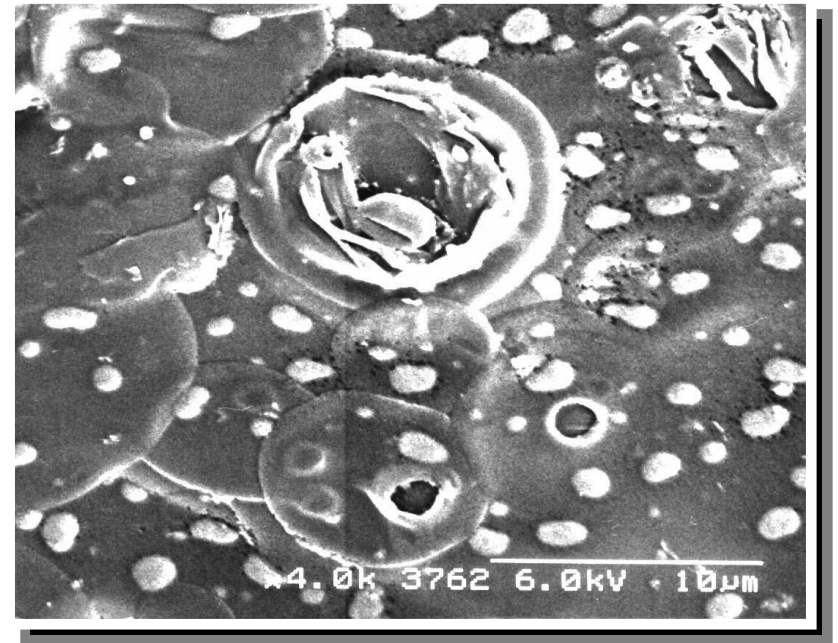
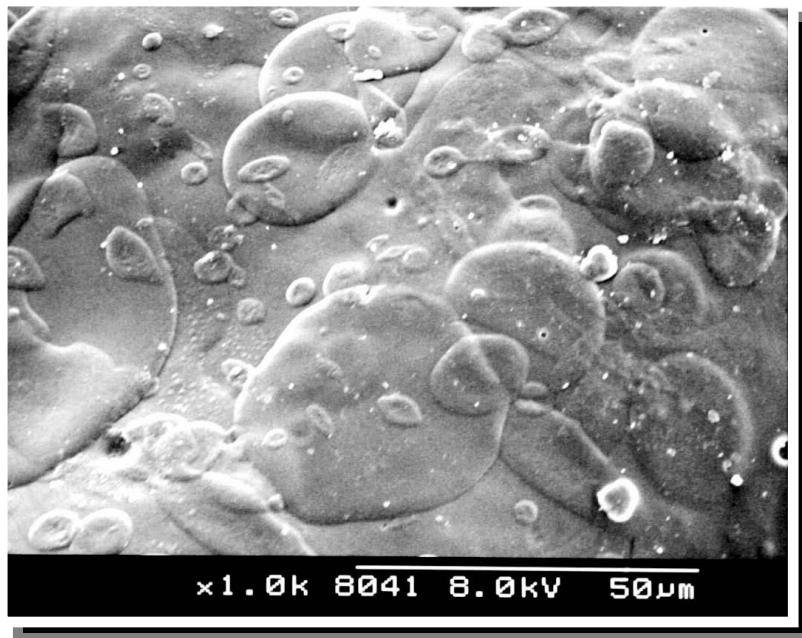


bridging



Coating defects

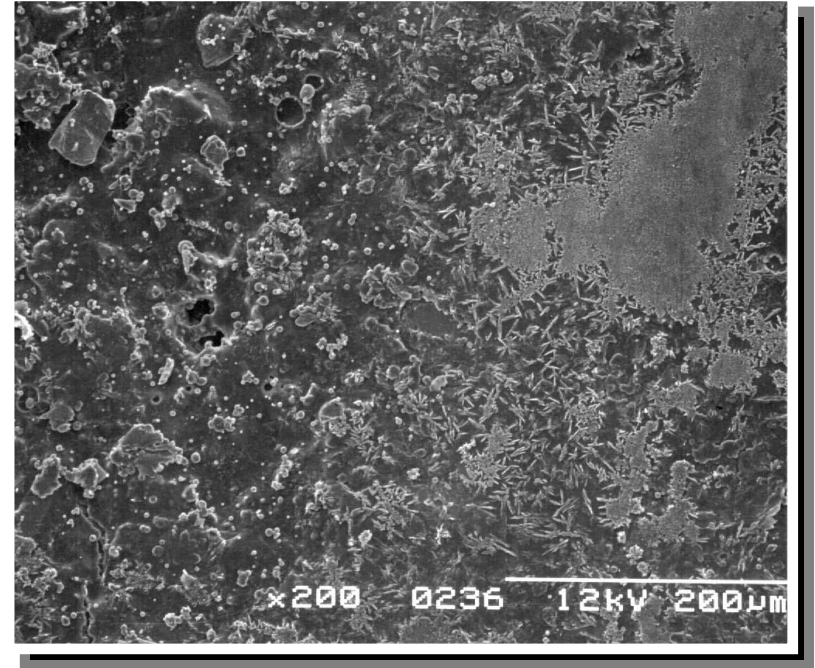
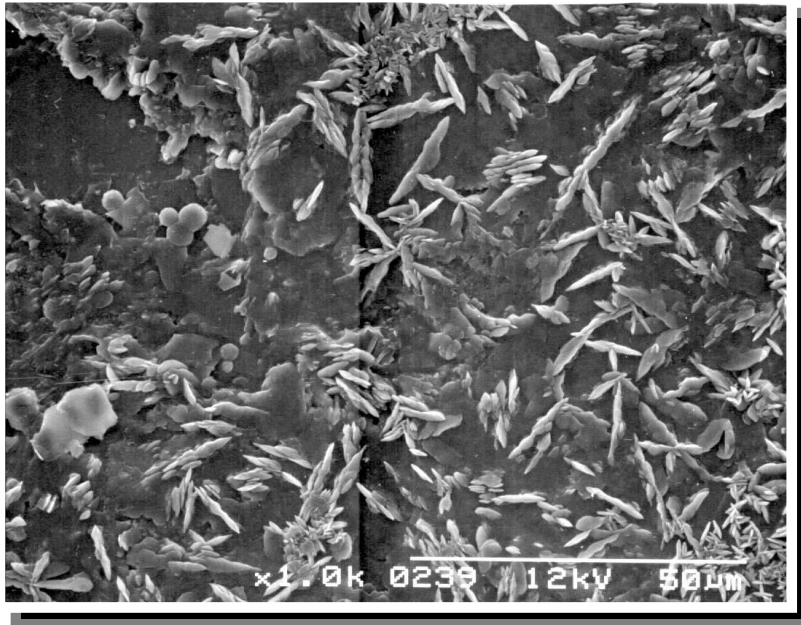
layering



bubbling

Coating defects

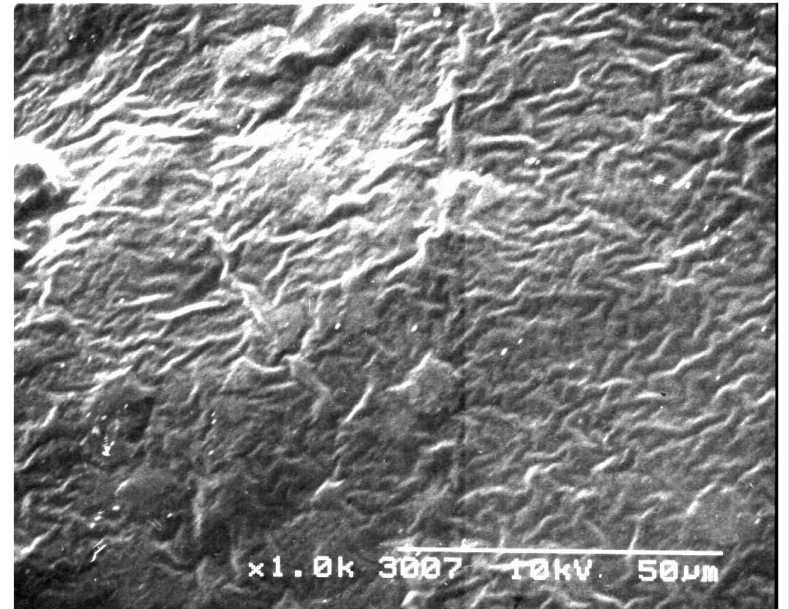
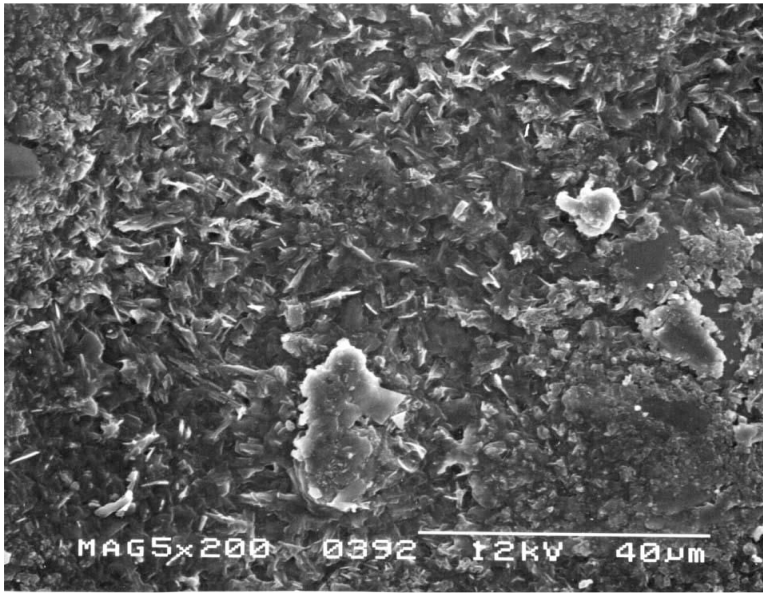
recrystallization



marbly

Coating defects

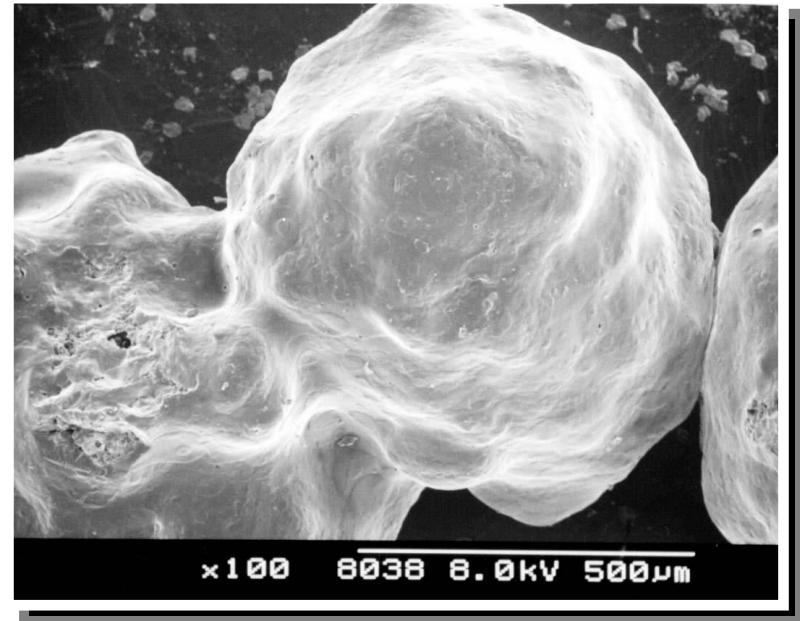
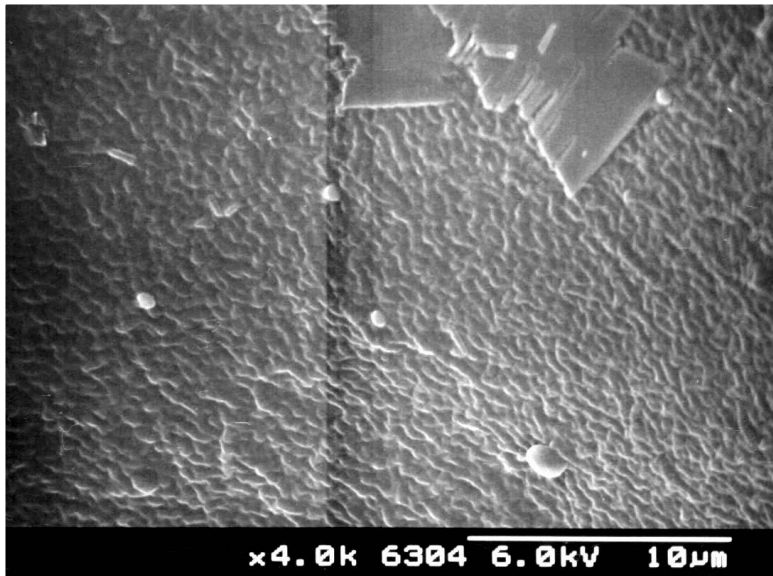
scaly



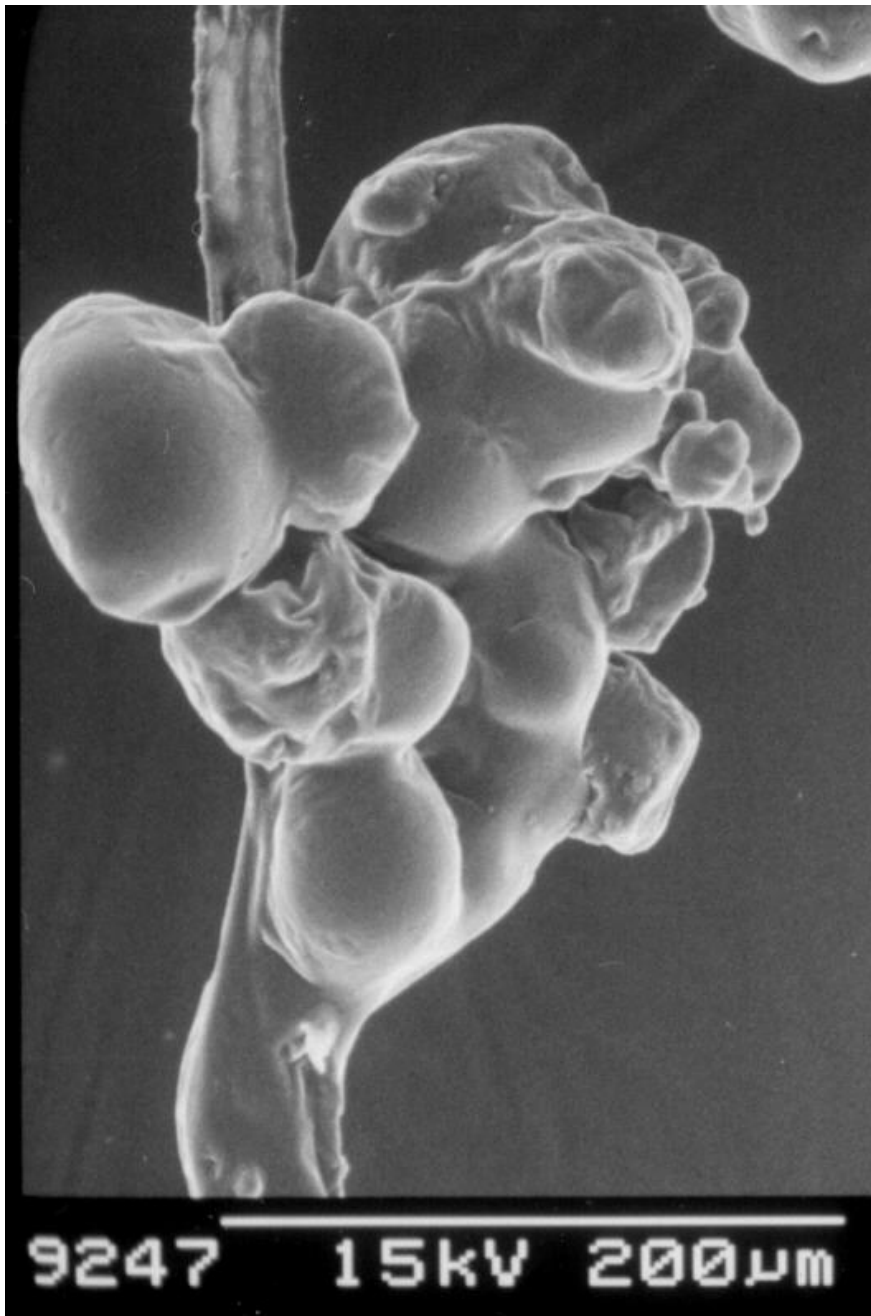
wrinkling

Coating defects

orange peel



twin formation



~~*Cluster of
grapes?*~~

It is the result of bad pellet coating.



Good bye!